



- 8 February 2013 Presqu'île de Giens - Hyères les Palmiers - France

Oxygenated volatile organic compounds (OVOCs) first detected from IASI: understanding the sources

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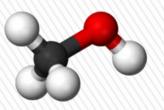






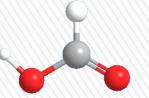
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CH₃OH

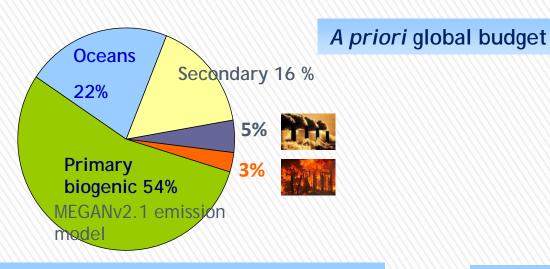


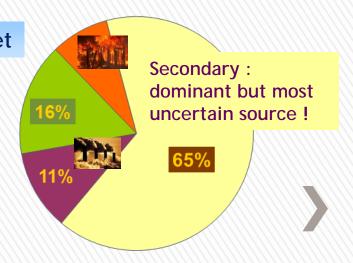
Abundant over forests in the growing season, impact on OH concentrations in the PBL

HCOOH



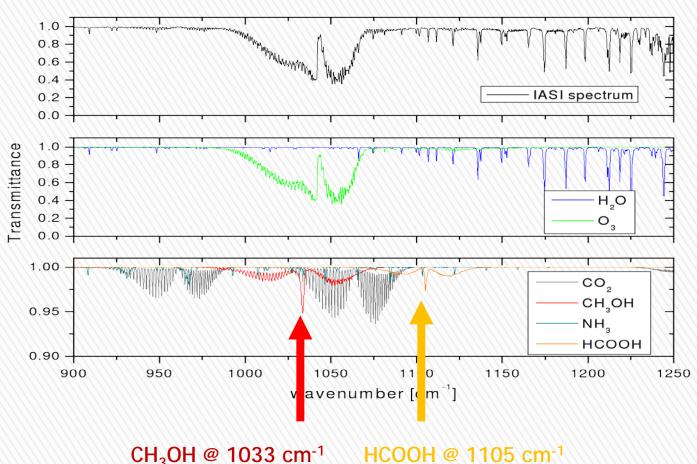
Major contributor to acidic rain in remote environments, but models underpredict severely the observed HCOOH concentrations → missing sources





Global source = 193 Tg/yr, T = 6 days

Global source = 36 Tg/yr, T = 4 days

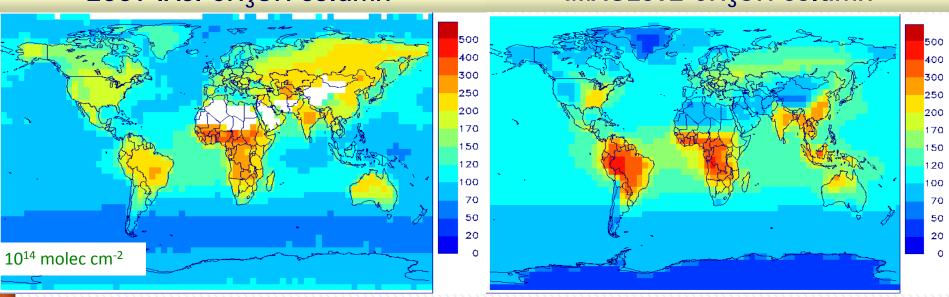


Very weak absorption interfered by other molecules, especially ozone

Retrieval based on $\Delta T_{\rm b}$ between the target channel and nearby reference channels in the spectrum baseline

2009 IASI CH₃OH column

IMAGESv2 CH₃OH column

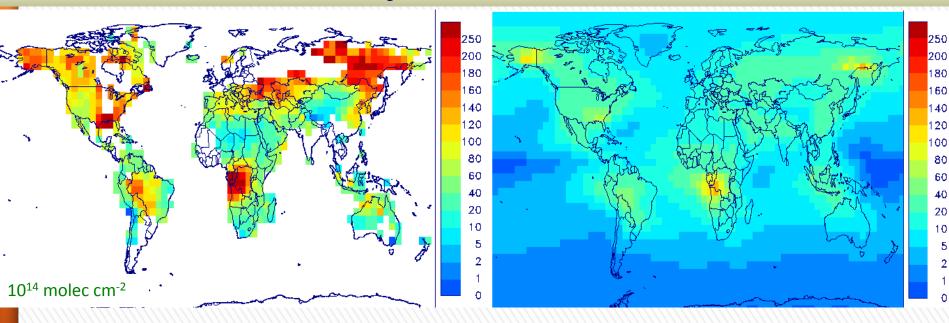


High columns over the majority of continental regions

Large overestimations over Amazonia (factor of 2-3), Africa &Indonesia (1.5-2), moderate over Europe, Eastern US

IASI HCOOH column - July

IMAGESv2 HCOOH column



Strongly enhanced columns over central Africa, S. America, and boreal regions

Important underestimations (factor of 2-3)

Source inversion using the adjoint model



Optimized emission distributions

$$G_0(x,t) = \sum_{j=1}^m \Phi_j(x,t)$$

$$G(x,t,f_j) = \sum_{j=1}^m \exp(f_j) \Phi_j(x,t)$$

Observation

Emission parameters f

Forward CTM

Transport Chemistry

Cost function J(f)

$$J(f) = \frac{1}{2} \left((H(f) - y)^T E^{-1} (H(f) - y) + (f - f_B)^T B^{-1} (f - f_B) \right)$$

Adjoint CTM

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Adjoint transport

Adjoint chemistry

Adjoint J(f)

Gradient of J(f)
Calculation of new f's
Is J(f) minimum?

no!

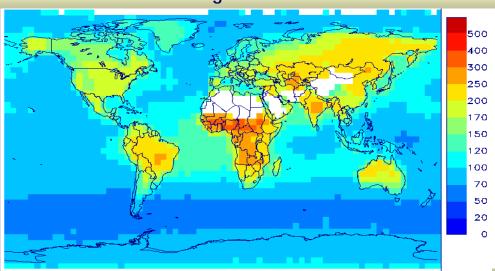
yes!

Handle very large numbers of control variables Address non-linearities

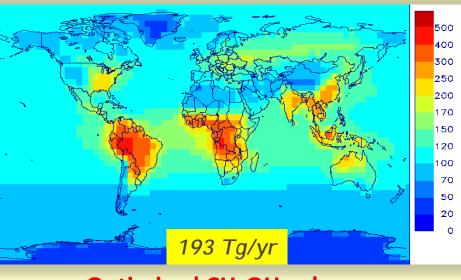
Distinguish between emission categories

Updated emission parameters

IASI CH₃OH column



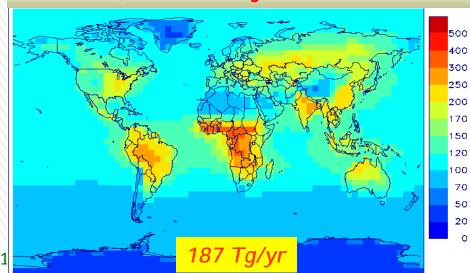
IMAGESv2 CH₃OH column



★ Global *a posteriori* biogenic CH₃OH source very close to the *a priori* MEGANv2.1

× BUT...

Optimized CH₃OH column

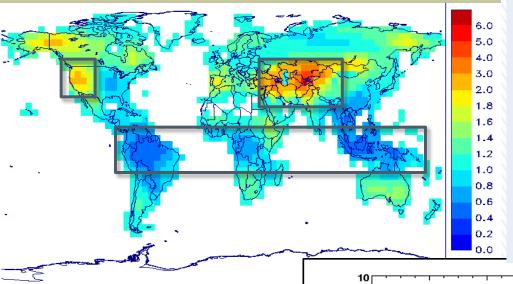


Stavrakou et al. ACP, 2011

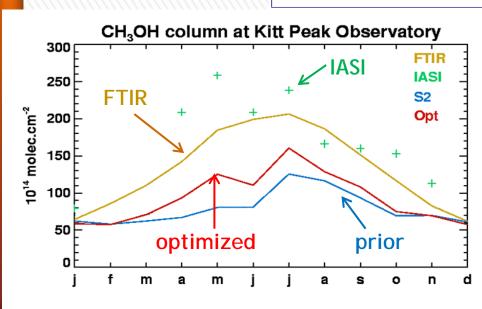
Strong flux increases in arid/semi-arid regions in central Asia (x5), western US (x2)

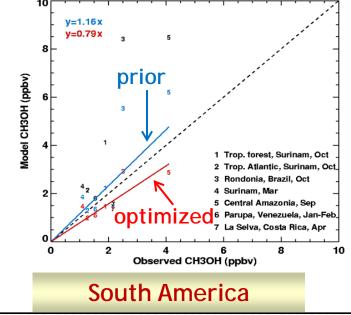
→ underestimated methanol rate from shrub or an unaccounted methanol source in MEGANv2.1?





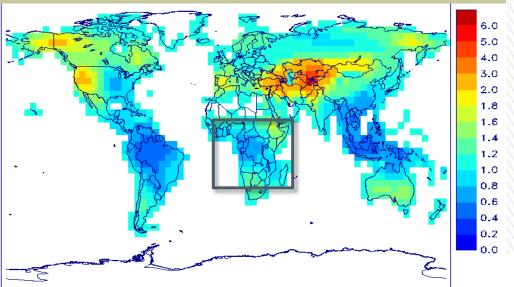
Significant reductions over the tropical forests in the Amazon (40-55%) and Indonesia (45-60%)





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Biogenic emission update

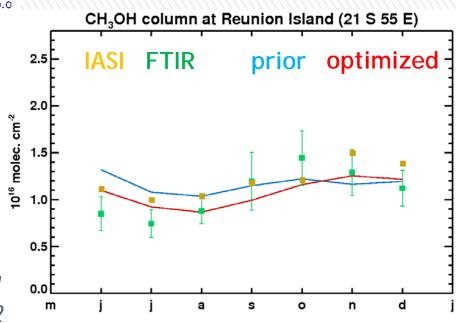


✓ FTIR supports IASI-derived decrease in biogenic emissions over Central & S. Africa

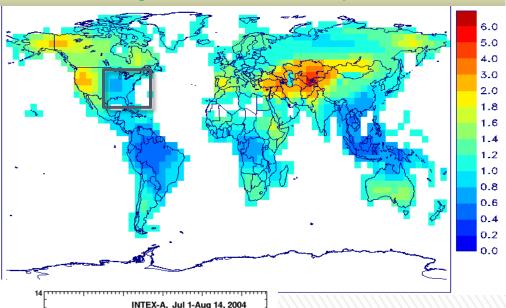


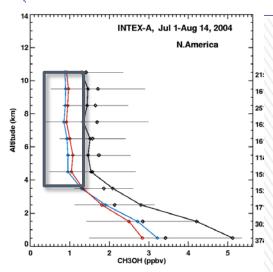
✓ Observed seasonality well reproduced by the model

Stavrakou et al. ACP, 2011 Vigouroux et al. ACP, 2012

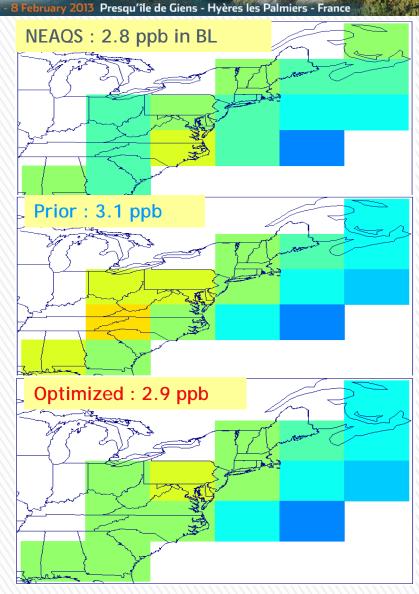


Biogenic emission update





Underestimation in the free troposphere: related to model underestimation in the Western US?

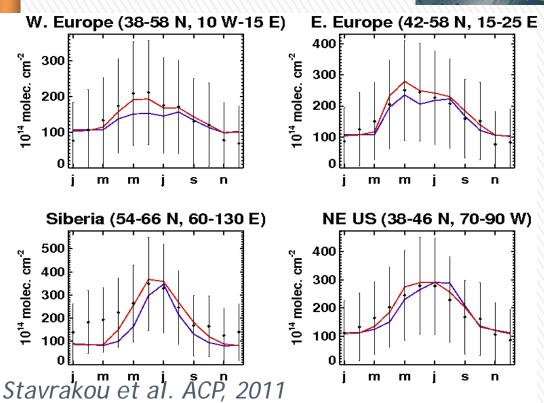


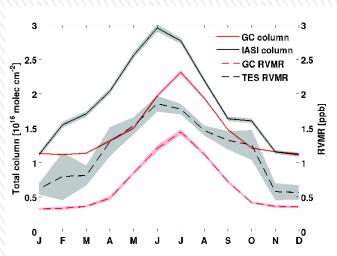
■ Very good agreement with aircraft data in PBL

INTEX-A (Jul 1-Aug 14 N. America)

Prior Optimized

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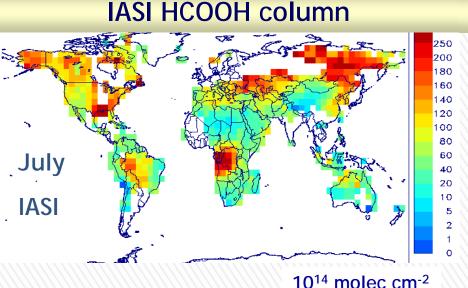
4. Seasonal cycle in atmospheric methanol over midlatifor 2009. Shown are methanol column amounts simulated EOS-Chem (base-case simulation, red solid line) and meaby IASI (black solid line), and representative volume mixing (RVMR) simulated by GEOS-Chem (base-case simulation,

Wells et al. ACP, 2012

- ✓ In mid-latitudes IASI is maximum in spring and early summer, i.e. 1-2 months earlier than the model peak, similarly for CH₃OH from TES (Wells et al., 2012) and tower measurements in the US Midwest (Hu et al., 2012)
- ✓ Emissions from new leaves are underestimated in MEGAN, and emissions from mature leaves should be decreased → important changes in leaf age classes for broadleaf trees

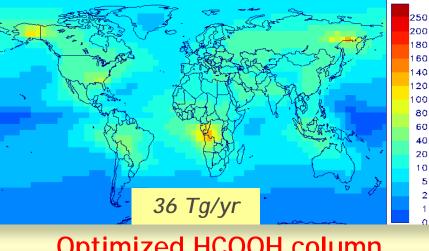
Optimize the global HCOOH source: assume that missing HCOOH is

produced through OH-oxidation of as-yet-unidentified biogenic precursors

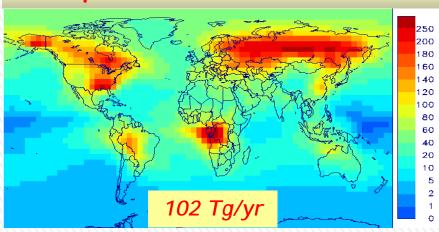


- Strong biogenic HCOOH source :
- ~ 3 x larger than in the a priori,
- ~90% of the global source
- Enhanced emission over boreal forests likely reflecting oxidation of BVOC from conifers





Optimized HCOOH column



10

300

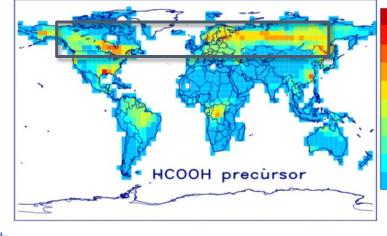
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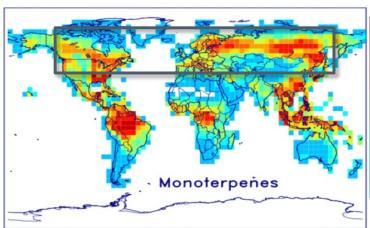
70

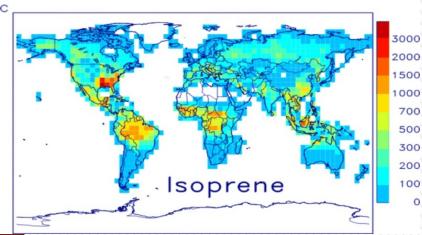
50 30

20

10



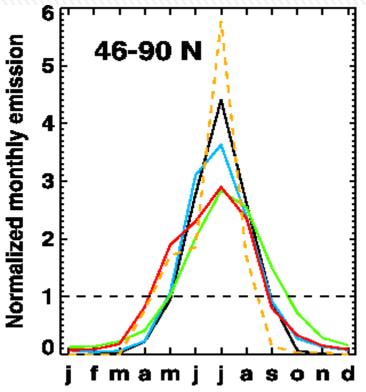




The distribution of the extra biogenic source suggests that oxidation of terpenoids leads to substantial amounts of HCOOH

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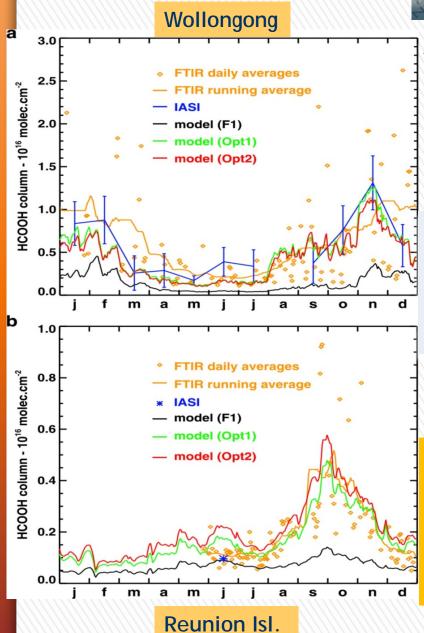
Should the entire missing source of HCOOH be due to monoterpenes only, a molar yield of 200% HCOOH would be required



- Isoprene (MEGAN)
 monoterpenes (MEGAN)
 formic acid (Lathiere)
- - blomass burning (GFED)
 - formlc acld precursor (this study)

- Summertime maximum
- Relatively high emissions also in spring at high latitudes, in agreement with field studies on monoterpene emissions
- Field measurements over boreal forests point towards the existence of large emission of undetected short-lived organic compounds (Di Carlo et al. Science, 2004)

Comparison with FTIR data





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✓ A priori mean underestimation by factor of 3-4 → excellent agreement after inversion

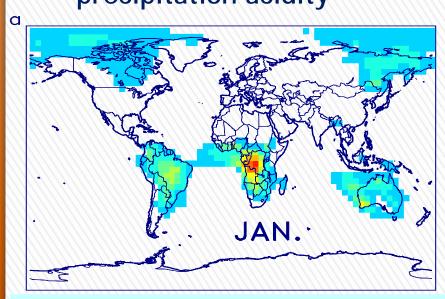
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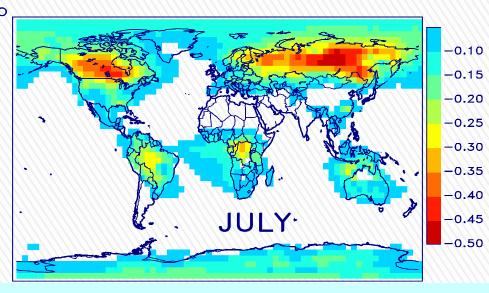
✓ Comparison with airborne data (INTEX-B)

Region	Observed	A priori	Optimized
Western US	490 ppt	66	230
North Pacific	140 ppt	28	120

✓ Comparison with ground-based data

нсоон	Observed	A priori	Optimized
Gas-phase (ppbv)	1.33	0.29	0.78-1.17
Rain (µmol/L)	6.7	1.32	4-4.2

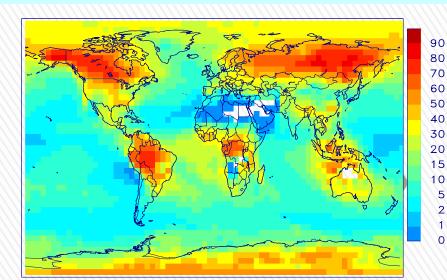




pH reduced by 0.25-0.5 over boreal forests in summer, 0.15-0.4 in the Tropics

Calculated contribution of HCOOH to the total concentration of [H+]

HCOOH accounts for as much as 60-80% of the rainwater acidity over Amazonia and boreal forests in summertime



Conclusions



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- Revisited the CH₃OH & HCOOH global distribution and budget
- ✓ CH₃OH: globally inferred biogenic source: ca. 100 Tg/yr, IASI suggests higher emissions over arid regions, and lower over tropical forests, supported by independent data, issues with seasonality in the mid-latitudes
- ✓ HCOOH: biogenic secondary source ~ 90% of the total source, originates mostly in boreal and tropical forests → high-yield product in the oxidation of organic compounds, such as monoterpenes and other terpenoids
- ✓ IASI measurements pave the way to a better understanding of the origin and fate of oxygenated compounds